Raman Spectroscopy: Introductory Tutorial

Daniel T. Schwartz

Department of Chemical Engineering Box 351750 University of Washington Seattle, WA 98195-1750

dts@u.washington.edu

Goal of the tutorial is to answer the questions,

"What is Raman spectroscopy and can the new Raman microscope at NTUF help with my sample?"



Raman Spectroscopy: Some Sources

General Principles and Instrumentation:

Principles of Instrumental Analysis, by Douglas A. Skoog, F. James Holler, Timothy A. Nieman

Inorganic:

Infrared and Raman Spectra of Inorganic and Coordination Compounds : Theory and Applications in Inorganic Chemistry (Volume A) by Kazuo Nakamoto

Infrared and Raman Spectra of Inorganic and Coordination Compounds : Applications in Coordination, Organometallic, and Bioinorganic Chemistry (Volume B) by Kazuo Nakamoto

Organic:

The Handbook of Infrared and Raman Characteristic Frequencies of Organic Molecules by Daimay Lin-Vien, et al

Raman Spectroscopy: Overview

- A vibrational spectroscopy
 - IR and Raman are the most common vibrational spectroscopies for assessing molecular motion and fingerprinting species
 - Based on **inelastic** scattering of a monochromatic excitation source
 - Routine energy range: 200 4000 cm⁻¹
- Complementary selection rules to IR spectroscopy
 - Selection rules dictate which molecular vibrations are probed
 - Some vibrational modes are both IR and Raman active
- Great for many real-world samples
 - Minimal sample preparation (gas, liquid, solid)
 - Compatible with wet samples and normal ambient
 - Achilles Heal is sample fluorescence

Raman Spectroscopy: General

• IR and Raman are both useful for Fingerprinting



• Symmetry dictates which are active in Raman and IR

Raman Spectroscopy: General

• Group assignments identify characteristic vibrational energy



Raman Spectroscopy: Classical Treatment

• Number of peaks related to degrees of freedom

DoF = 3N - 6 (bent) or 3N - 5 (linear) for N atoms

• Energy related to harmonic oscillator

• Selection rules related to symmetry

Rule of thumb: symmetric=Raman active, asymmetric=IR active

CO_2	H ₂ O
$ \begin{array}{c} & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & $	Raman + IR: 3657 cm^{-1} Raman + IR: 3756 cm^{-1} Raman + IR: 1594 cm^{-1}
$+ \omega = \omega +$	$\rightarrow \checkmark$

Main Optical Transitions: Absorption, Scattering, and Fluorescence



Raman Spectroscopy: Absorption, Scattering, and Fluorescence



Raman Spectroscopy: At NTUF, you pick the Laser Excitation



Raman Spectroscopy: Absorption, Scattering, and Fluorescence



Raman Spectroscopy: Coping w/ Fluorescence

- 1. Use NTUF 785 nm laser line (excites many fewer fluorophores)
- 2. Photobleach with long exposure laser irradiation.



Raman Spectroscopy: Summary

- 1. Raman is a vibrational spectroscopy akin to IR
 - Good for fingerprinting, probing molecular symmetry
- 2. Scattering-based, not transmission/reflection
 - Means no need for fancy sample preparation...gas, liquid, or solid
 - Virtually always use anti-Stokes lines due to stronger signal
- 3. You need to pick excitation energy (laser line)
 - 785 nm: Fluorescence less probable; Lower Raman signal
 - 514 nm: Fluorescence more probable; Resonance more likely; Higher signal
- 4. Other things not talked about
 - SERS: Surface Enhanced Raman Spectroscopy
 - Quantum origins of selection rules and scattering cross-section

Fuel Cell Materials --- picked because I had little Raman experience with these materials



From http://www.fueleconomy.gov/feg/fc_pics/fuel_cell_still.gif

Flow Field Plate - Graphite $\lambda_{ex} = 514.5 \text{ nm}$ Intensity (counts) $\left(\frac{I_g}{I_s}\right) = 3.98$ 500 2000 2500 1000 1500 0 Raman Shift, $\Delta \sigma$ (cm⁻¹)



Nanocrystalline graphite has graphitic (g) and disorder (d) peaks. The characteristic dimension of graphitic domains is given by:

$$| I(\mathring{A}) = 44 \left(\frac{I_g}{I_d} \right) = 175 \mathring{A}$$

From early literature on graphitic materials Tuinstra and Koenig, J. Chem Phys. 53, 1126 (1970).







