

## Homework #1 Chem 524 2011

To be handed in when requested, as problems build up

[Homework link to set #1 \*Assignment – Notes 1 \(background\)\*](#):

**Read in Text:** this section covers: **Chapter 1, Chapter 2-2, 2-3, 2-5**

**Study on your own: Chap.2-4** Selection of information (we will do later)

**Chap.2-5** Analytical signal — Sorting out various contributions to the signal measured from background, etc. — This topic will recur and be tested.

**Homework:** (For discussion only: Chap. 1: #3, 4, 6, 10, Chap. 2: #5, 10;  
Chap2: #1, **good model test question**, previously on 1<sup>st</sup> exam)  
You should already know how to solve/use: Chap 1: #1,12, Chap 2:  
#6,13, 14

**To eventually hand in:** Chap.2: # 4, 7, 11, 15, plus

- Consider H<sub>2</sub>O, calculate the population of the  $v = 0$  and 1 molecular states for the vibrational states corresponding to  $\nu = 3600$  and  $1650 \text{ cm}^{-1}$  for  $T = 1000\text{K}$  and  $300\text{K}$  (room temperature). Do the same for the  $J = 1, 2$  and 3 rotational states of HCl, assuming  $B$  is  $\sim 20 \text{ cm}^{-1}$ .
- I have a glower in my IR spectrometer with a heated area roughly  $3 \times 5 \text{ mm}$ . The lamp is rated at  $50 \text{ W}$ . If its color temperature is  $1500\text{K}$ , calculate the wavelength of maximum intensity, assuming blackbody character, and then determine the radiance assuming that 70% of the power comes out as light and that you can use  $\lambda_{\text{max}}$  as an average wavelength/frequency mark.

**Homework for Notes 2 (incoherent sources):**

**Read Chapter 4 Section 1,2**, also look at *sources tutorial on Oriol/Newport* site

Problems in the book:

For discussion: Ch. 4 - #12,

**To hand in eventually: Ch. 4 - # 1**

## Homework for Notes 3 (laser sources)

### 3. Laser light sources:

Text reading this section covers: **Chapter 4-3** – pretty inadequate, out of date

Also review [Kansas State web pages provided in links](#),

<http://www.phys.ksu.edu/perg/vqm/laserweb/Preface/Toc.htm>

Problems in the book - For discussion Ch. 4-18 *and*

Consider best choice laser sources for the following, rationalize your selection:

- Raman spectrometer, routine with microscope for materials
- Resonance Raman spectrometer for small molecules
- T-jump fluorimeter for biological systems, like proteins
- 2D IR correlation IR of fs pulses,
- Very high resolution IR of gases for pollution detection
- laser ablation/ pulsed beam measurements
- MPI molecular beam studies of small molecules

**To hand in eventually: textbook: Ch. 4 - # 2,14** and *a and b* below:

**a.** from O. Svelto and D.C. Hanna, *Principles of Lasers, 2nd Ed*, Plenum, 1982.

1.4 If two levels at 300° K are in thermal equilibrium with  $n_2/n_1 = 0.2$ , calculate the frequency of the transition from  $1 \rightarrow 2$ . In what part of the spectrum does this occur? Change this to 0.005 and recalculate.

2.0 Calculate the number of longitudinal modes that occur in  $\Delta\nu = 1 \text{ cm}^{-1}$  at  $\lambda_0 = 488 \text{ nm}$  for a 0.7 m long laser cavity.

1.6. Ultimate limit of divergence of a laser is diffraction  $\theta_d = \beta\lambda/D$  where

$\theta_d$  = divergence,  $\lambda$  = wavelength,  $\beta \sim 1$  optimal design,  $D$  = diameter

If a YAG:Nd laser beam ( $\lambda = 1.06 \mu$ ) is sent to the moon (384,000 km) from an oscillator of  $D = 1 \text{ mm}$ , calculate its diameter on arrival.

**b.** from Kansas State site **Question 4.4: Ar+ Ion laser**

The difference between adjacent modes in Ar+ Ion laser is 100 MHz. The mirrors are at the end of the laser tube. **Calculate:**

- The length of the laser cavity.
- The mode number of the wavelength 488 [nm].
- The change in separation  $\Delta\lambda$  of adjacent modes when the cavity is shortened to half its length.